

# Read PDF Naming Hydrates Answers

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## **Nomenclature Ans for Prep Chem - CCBC Faculty Web**

The “n” is usually a low integer though it is possible for fractional values to exist. The prefixes are the same Greek

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prefixes used in naming molecular compounds. Therefore, in a monohydrate “n” is one; in a hexahydrate “n” is 6, and so on. The Greek prefixes used in naming hydrates for numbers 1/2 through 10 are as follows: 1/2: hemi-

## **LAB % COMPOSITION OF A HYDRATE**

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## **LAB**

The purpose of this lab is to determine the number of water molecules associated with unknown hydrated salts. Closed captions available. 00:30 salt definitio...

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Organic Hydrates. In organic chemistry, hydrates tend to be rarer. An organic hydrate is a compound formed by the addition of water or its elements to another molecule. For example, ethanol,  $\text{CH}_3\text{-CH}_2\text{-OH}$ , can be considered a hydrate of ethene,  $\text{CH}_2=\text{CH}_2$ , formed by the addition of H to one C and

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## **ChemTeam: Hydrates**

Naming practice: hydrates Write the formula for the following hydrates: 1) zinc sulphate heptahydrate 2) copper (I) sulphite monohydrate 3) cobalt (II) fluoride tetrahydrate 4) lithium nitrate trihydrate 5) sodium sulphate decahydrate Write the compound name for the following formulas: 1)  $\text{Na}_2\text{SO}_4 \cdot \text{H}_2\text{O}$  \*



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1)  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  2)  $\text{LiNO}_3 \cdot 3\text{H}_2\text{O}$  3)  $\text{CuSO}_4 \cdot 3\text{H}_2\text{O}$   
4)  $\text{Ca}(\text{NO}_3)_2 \cdot 3\text{H}_2\text{O}$  5)  $\text{NaClO}_3 \cdot \text{H}_2\text{O}$

## **Naming Compounds | Boundless Chemistry**

LAB % COMPOSITION OF A  
HYDRATE.docx. At the end of the lab is  
an example page of how to do the  
calculations on page 22 after you have

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completed the data. Print out this page also. Chemistry: Lab - % Composition of a Hydrate Introduction: Salt Compounds that contain water molecules attached are called hydrates, or hydrated salts. Salt ...

**ANSWER KEY\_Worksheet - Naming Ionic Compounds and Hydrates ...**

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Learn naming acids hydrates with free interactive flashcards. Choose from 500 different sets of naming acids hydrates flashcards on Quizlet.

## **chemistry help...naming hydrates and ... - Yahoo Answers**

What are hydrates? If you're studying chemistry, you probably need to know

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what a hydrate is and the role it serves. In this guide, we give a complete explanation of hydrates, including the hydrate definition, the three different types of hydrates, the rules you need to know to name hydrates and write out their formulas, and common hydrate examples you may have heard of.

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## **Hydrate Worksheet - Council Rock School District**

WORKSHEET #7: NOMENCLATURE OF  
HYDRATED IONIC COMPOUNDS Use Latin  
prefixes to indicate # of water molecules  
present 1 = mono 2 = di 3 = tri 4 =  
tetra 5 = penta 6 = hexa 7 = hepta 8 =  
octa 9 = nona 10 = deca # Name of  
Hydrate Chemical Formula Eg.

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copper(II)sulfate pentahydrate  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$   
1 magnesium sulfate heptahydrate  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$   
2 sodium carbonate decahydrate  $\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$   
3

## **Quiz: Naming Acids, Hydrates, and Covalent Compounds ...**

Problem #3: Name this substance:  $\text{MgSO}_4 \cdot \text{H}_2\text{O}$   
magnesium sulfate monohydrate. When there is only one

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water, make sure to still use the proper prefix. Magnesium sulfate hydrate is not the correct name for  $\text{MgSO}_4 \cdot \text{H}_2\text{O}$ . .  
By the way, multiple hydrates exist for  $\text{MgSO}_4$ . In addition to the monohydrate, there also exists the tetrahydrate, pentahydrate, hexahydrate and heptahydrate.

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## **Worksheet - Naming Ionic Compounds and Hydrates - Answer**

...

Test your knowledge with the Naming Acids, Hydrates, and Covalent Compounds Naming Quiz quiz.

**Naming Hydrates - sartep.com**  
for the following hydrates: Barium



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Phosphate dihydrate. Copper(II)sulfate pentahydrate. Cobalt chloride hexahydrate. Magnesium sulfate heptahydrate. Nickel (II) sulfate hexahydrate. Copper (I) Bromide tetrahydrate. Naming Hydrates. Write the correct numerical prefix for each of the numbers in the table. Number Prefix

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## **Naming Acids, Hydrates, and Covalent Compounds Naming Quiz**

...

Nomenclature of Hydrated Ionic Compounds. Select your answer by clicking on the appropriate button (buttons are labeled with a question mark, "?"). Rules for Naming Hydrated Ionic Compounds. Hydrated ionic

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compounds (i.e., hydrates) have a specific number of water molecules in their chemical formulas. In the solid, these water molecules (also called "waters of hydration") are part of the ...

## **Naming Hydrates | Introduction to Chemistry**

Start studying Naming Acids, Hydrates,

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and Covalent Compounds Naming Quiz. Learn vocabulary, terms, and more with flashcards, games, and other study tools.

## **naming acids hydrates Flashcards and Study Sets | Quizlet**

Naming a Hydrate Hydrates are compounds that attract and bond with

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water molecules when they crystallize. The two parts of a hydrate are the anhydrous salt and the water of hydration. The ratio of these water molecules in the compound is always a whole number.

## **Naming Ternary Acids and Hydrates - Gateway School District**

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Note that some of the cations required Roman numerals, and some do not! Remember that Ni  $2+$  and Zn  $2+$  are two of the unusual cations that have fixed charges even though they are not from the Group IA, IIA families. You should review which other cations are in this category. Note that some of the cations required Roman numerals, and

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some do not!

## **What Are Hydrates? Definition, Naming, and Examples ...**

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KEY\_Worksheet - Naming Ionic  
Compounds and Hydrates.pdf from CH  
204 at University of Texas. Naming Ionic  
Compounds and Hydrates Answer Key

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Write the name for each of the

## **111L Hydrates (#4)**

Hydrate, any compound containing water in the form of H<sub>2</sub>O molecules, usually, but not always, with a definite content of water by weight. The best-known hydrates are crystalline solids that lose their fundamental structures



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upon removal of the bound water.  
Exceptions to this are the zeolites

## **WORKSHEET #7: NOMENCLATURE OF HYDRATED IONIC COMPOUNDS**

...

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Naming Ionic Compounds and Hydrates -  
Answer Key from CH 204 at University of

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Texas. Naming Ionic Compounds and Hydrates Answer Key Write the name for each of the

## **HELP WITH CHEMISTRY QUESTION? | Yahoo Answers**

Naming Ternary Acids and Hydrates  
Name\_\_\_\_\_ Directions: Answer the following questions about naming

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ternary acids and hydrates. 1. Epsom salts has the formula  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ . What is the chemical name of this

## **Nomenclature of Hydrated Ionic Compounds**

nomenclature is essential in chemistry and hydrates follow a lot of rules of ionic compounds (thankfully) you first name

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the ionic compound normally (cation, then anion) then you name the water part of the hydrate using the latin prefixes for molecular compounds. and adding the word "hydrate" mono 1. di 2. tri 3. tetra 4 . penta 5. hexa 6 ...

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